

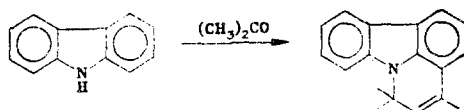
CONDENSATION OF CARBAZOLE WITH ACETONE IN THE PRESENCE OF
TRIFLUOROACETIC ANHYDRIDE. A SINGLE STAGE SYNTHESIS OF
1,1,3-TRIMETHYL-1H-PYRIDO[j,k]CARBAZOLE

N. V. Moskalev and V. D. Filimonov

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The acid catalyzed reaction of carbazole with aliphatic aldehydes has been well investigated and can lead to oligomers of poly-9-(alken-1-yl)carbazoles or 9,9'-dicarbazolyldialkyl ethers [1, 2] or to 9-(1-alkoxyalkyl)carbazoles [3] in the presence of alcohols. Attempts to involve ketones in these reactions have been unsuccessful up to this time.

We now report the first example of the reaction of carbazole with a ketone under mild conditions. In the presence of trifluoroacetic anhydride carbazole readily condensed with acetone to form 1,1,3-trimethyl-1H-pyrido[j,k]carbazole:



In the ¹³C NMR spectrum the signal for the quaternary C₁ was found at comparatively low field (55.8 ppm) which confirmed the proposed structure and excluded an alternative double bond isomer position.

1,1,3-Trimethyl-1H-pyrido[j,k]carbazole. Trifluoroacetic anhydride (20 ml, 143 mmole) was added slowly (10 h) with stirring to a solution of carbazole (5 g, 30 mmole) in acetone (90 ml, 1.4 mole) and the stirring continued for a further 10-12 h. The product was diluted with benzene (20 ml), washed with water (3 × 300 ml), the organic layer removed and dried with CaCl₂. The solvent was distilled off and the oily residue dissolved in heptane (7 ml) and the product mixture separated on an Al₂O₃ column (3 × 60 cm. activity II) using heptane as eluent. 1,1,3-Trimethyl-1H-pyrido[j,k]carbazole (3.6 g, 66%) was obtained with mp 113-114.5°C (from heptane) and R_f 0.5 (benzene:hexane 1:1), IR Spectrum (Nujol): 1620, 1590 (C=C), 835, 755 cm⁻¹ (C-H). It was strongly triboluminescent in the visible region. PMR Spectrum (CCl₄): 1.70 (6H, s, 1-CH₃), 2.05 (3H, s, 3-CH₃), 5.22 (kH, s, =CH), 6.8-7.3 (4H, m, 5-, 8-, 9-, and 10-H), 7.55 (1H, d, J_{4,5} = 7 Hz, 4-H), 7.65 (1H, d, J_{7,8} = 7 Hz, 7-H), 7.90 ppm (1H, d, J_{5,6} = 7 Hz, 6-H).

Elemental analytical data agreed with that calculated.

LITERATURE CITED

1. E. E. Sirotkina, V. D. Filimonov, N. A. Tsekhanovskaya, and L. S. Sizova, US Patent 3,987,011; Chem. Abstr., **86**, 30298 (1977).
2. E. O. Koval', Manuscript Deposition, VINITI No. 673-B87; Russ. Zh. Khim., No. 11, Zh 254 (1987).
3. V. A. Anfinogenov, V. D. Filimonov, and E. E. Sirotkina, Zh. Org. Khim., **14**, 1723 (1978).

S. M. Kirov Polytechnic Institute, Tomsk 634004. Translated from Khimiya Geterotsiklicheskich Soedinenii, No. 12, pp. 1694-1695, December, 1987. Original article submitted June 30, 1987.